

# Chapter 23

\* Ray model of light: light travels in straight lines in uniform transparent media (like air and glass)   
 الاشعة تاتي بتقني في خط مستقيم في نفس الوسط.

Geometric optics: light rays travel in straight lines at different angles.

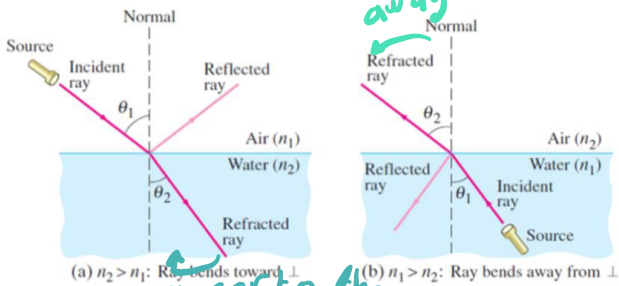
Index of refraction:

Light has the maximum speed in vacuum  $c = 3 \times 10^8$  m/s

$$n = \frac{c}{v} \rightarrow \begin{matrix} c > v \\ n > 1 \\ n = 1 \text{ Vacuum} \end{matrix}$$

انكسار

Refraction: "Snell's Law"



closer to the normal

$$n_2 > n_1$$

\* air  $\rightarrow$  water

$$\theta_2 < \theta_1$$

it will bend towards

$$n_2 < n_1$$

water  $\rightarrow$  air

$$\theta_2 > \theta_1$$

it will bend away

From the normal.

Snell's Law:

$$n_1 \sin \theta_1 = n_2 \sin \theta_2$$

$$\sin \theta_c = \frac{n_2}{n_1}$$

$$\text{if } n_1 < n_2 \rightarrow \theta_1 > \theta_2$$

$$\text{if } n_1 > n_2 \rightarrow \theta_1 < \theta_2$$

Note:

• معامل الانكسار  $n$  لكل الاوساط دائراً اكبر من 1 لانه سرعة الضوء بالفراغ اسرع من

• In Vacuum:  $n = 1$  the lowest  $c = 3 \times 10^8$  the highest

• زاوية السقوط تساوي زاوية الانعكاس

انعكاس كلي لا يوجد انكسار

انعكاس كلي

● Total internal reflection: "Fibre optics"

- \* when light passes from a medium of high  $n$  to a lower  $n$ .
- \*  $\theta_c$ : critical angle: Refracted ray travels parallel to the surface  $\theta_2 = 90$
- \*  $\theta_i > \theta_c$ : No light is refracted and all light is reflected. كل الشعاع ينعكس ما ينكسر
- \* Fibre optics: Medical instrument
- \* Glass and plastic has a few ( $10^4$  m)

● Thin Lenses, Ray Tracing:

Converging lenses عدسة محدبة

Diverging lenses عدسة مقعرة

Converging

Concave

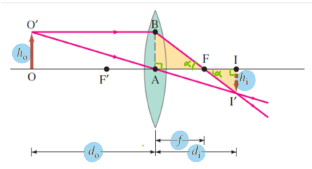
\* We call the lens is thin when the diameter of the lens is smaller than the radius of

Lens Power:

$P = \frac{1}{f}$  [P] =  $m^{-1}$  = diopter

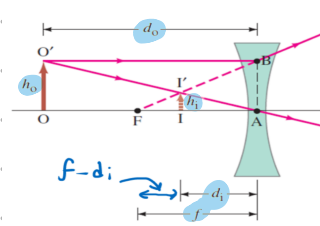
● Thin Lens Equation:

Converging lens:



+ for real obj.  
- for virtual obj.

Diverging lens:



$\frac{1}{f} = \frac{1}{d_o} + \frac{1}{d_i}$

$\frac{1}{f} = \frac{1}{d_o} + \frac{1}{d_i}$

f +  
 $d_o$  + real obj  
 $d_i$  <  $d_o$  > f  
 <  $d_o$  < f

$h_o$  +  
 $h_i$  <  $d_o$  < f  
 >  $d_o$  > f

f -  
 $d_o$  + real obj  
 $d_i$  - virtual image

$h_o$  +  
 $h_i$  +

## • Magnification (m):

$$m = \frac{h_i}{h_o} = -\frac{d_i}{d_o}$$

$m +$  → upright → real im.

$m -$  → inverted → virtual im.

- diverging lens can't form magnified images

## Note:

- a diverging lens always produces upright virtual image regardless the value of  $d_o$

- Converging lens:

real inverted images

virtual upright images

# Chapter 30

## Structure and Properties of the Nucleus:

Atom made up of electrons orbiting the nucleus

\* The nucleus is made up of protons and neutrons.

\* neutrons and protons are called nucleons.

\* Different nuclei are referred to as nucleus

Mass number: the sum of proton and neutron numbers in nucleus

$$A = Z + N$$

$Z$   $N$ : neutron number

Atomic number:  
number of protons  
in nucleus

\* The radius of nucleus:

$$R = 1.2 \times 10^{-15} A^{1/3}$$

$$1 \text{ fermi} = 10^{-15} \text{ m}$$

mass number  
 $A = Z + N$

\* atomic mass unit (u)

on this scale  $^{12}\text{C}$  has mass of 12u

\* Mass can also be measured in units of energy:

Einstein's equation:

$$E = mc^2$$

$$\frac{N}{N_0} = \left(\frac{1}{2}\right)^n$$

## Radioactivity:

$$N = N_0 e^{-\lambda t}$$

$N$ : remaining number of radioactive nuclei at time  $t$

$N_0$ : initial number of radioactive nuclei at time  $t_0 = 0$

$\lambda$ : decay constant  $\frac{1}{\text{time}}$

## Note:

- $P_{\text{nuc}} \Rightarrow P_{\text{atom}}$
- $1 \text{ eV} = 1.6022 \times 10^{-19} \text{ J}$

\* **half-life**: the time needed for the number of radioactive nuclei to become half the original number

$$t_{1/2} = \frac{\ln 2}{\lambda}$$

large  $\lambda \rightarrow$  small  $t_{1/2} \rightarrow$  fast decay  
small  $\lambda \rightarrow$  large  $t_{1/2} \rightarrow$  slow decay

Activity (A): number of decay per second

$$[A] = \text{decay/s} = \text{s}^{-1} / \text{Bq}$$

$$1 \text{ Ci} = 3.7 \times 10^{10} \text{ Bq}$$

$$A = A_0 e^{-\lambda t}$$

$$A = N \lambda$$

Becquerel

Mean life ( $\tau$ ): average life time of all the radioactive nuclei of a given radioactive element.

$$\tau = \frac{1}{\lambda} = \frac{t_{1/2}}{\ln 2}$$

# Chapter 31

## Absorbed Dose (AD):

The energy deposited per kg in any medium by any radiation type

\* Units:

1 Gray  $\rightarrow$  1 J/kg

1 rad  $\rightarrow$  0.01 J/kg

1 Gy = 100 rad

$$AD = \frac{\text{Energy}}{\text{mass}}$$

## Effective Dose (ED)

$$ED = AD \times RBE$$

## Note:

- $\alpha$  is the most dangerous radiation
- X-ray &  $\beta$  are the least dangerous radiation
- RBE has no unit

\* RBE: relative biological effectiveness: the number of rads of X-ray or  $\gamma$  that produces to the biological damage as one rad of given radiation

Radiation Type	RBE
X and $\gamma$ rays	1
$\beta$ (electrons)	1
Protons	2
Slow neutrons	5
Fast neutrons	$\approx$ 10
$\alpha$ particles and heavy ions	$\approx$ 20

the larger RBE the more dangerous

[AD] [ED]

Gray (Gy) Sievert

Rad rem

1 Gy = 100 rad

1 Sv = 100 rem

$$\frac{E_{\text{worker}}}{E_{\text{total}}} = \frac{A_{\text{worker}}}{A_{\text{total}}}$$

$$\text{Intensity of radiation} = \frac{\text{Energy}}{\text{unit area}}$$

$4\pi r^2$

بعد worker من المصدر

$E_{\text{total}} = E_{\gamma} A \rightarrow \text{J/s}$

Done

By

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