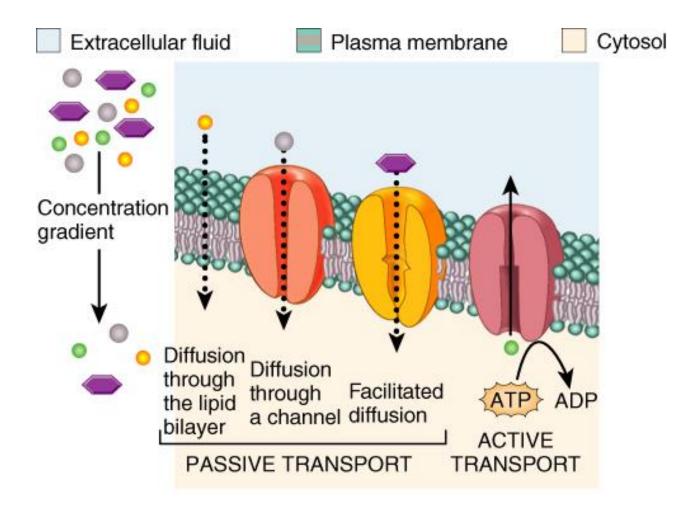
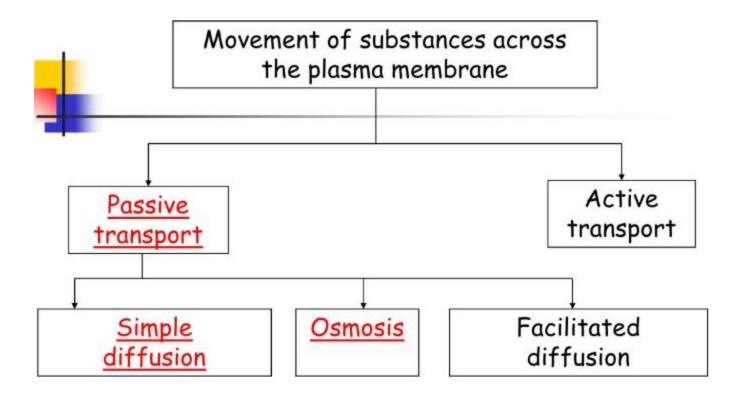
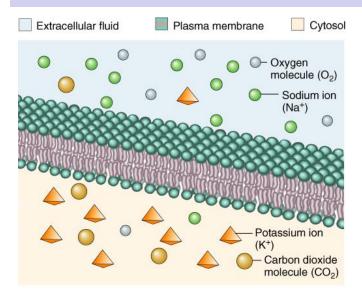
https://www.youtube.com/watch?v=A9ihz5gYxU4

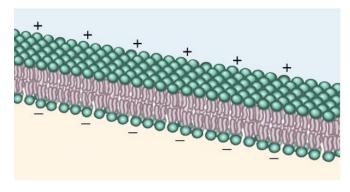




Diffusion through lipid bilayer



(a) Concentration gradients



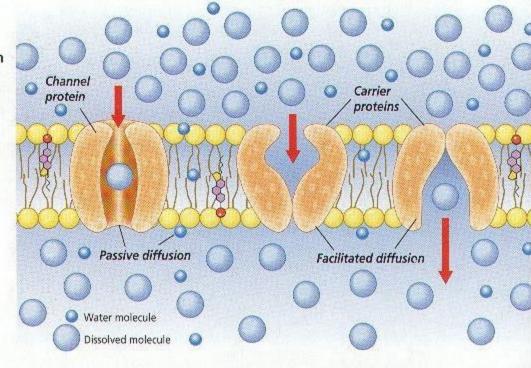
• CO2

- 02
- NO
- Steroid Hormones
- Monoglycerides

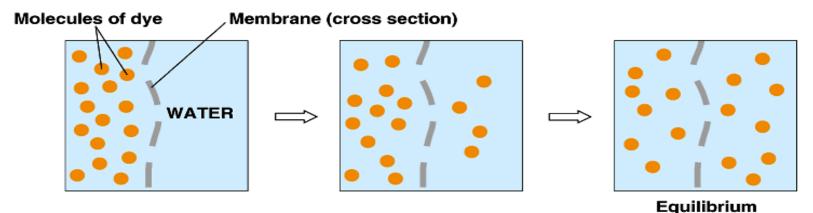
(b) Electrical gradient

Diffusion through channels

Channel proteins provide the openings through which small, dissolved particles, especially ions, diffuse by passive transport.

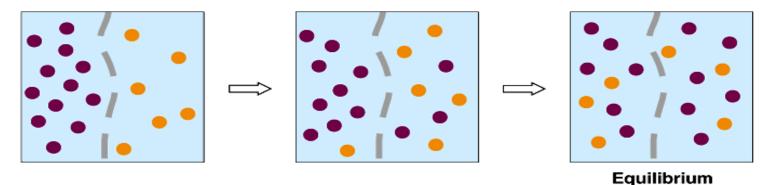


The Concept of Simple Diffusion



Equilibr

(a) Diffusion of one solute



(b) Diffusion of two solutes

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Are we needing to consume macro-energetic molecules (ATP) for diffusion across plasma membranes??

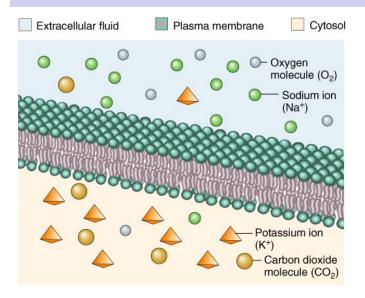
What type of Energy is there to get the diffusion??



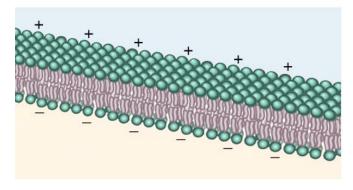
Figure 4-3

Diffusion of a fluid molecule during a thousandth of a second.

Diffusion through lipid bilayer



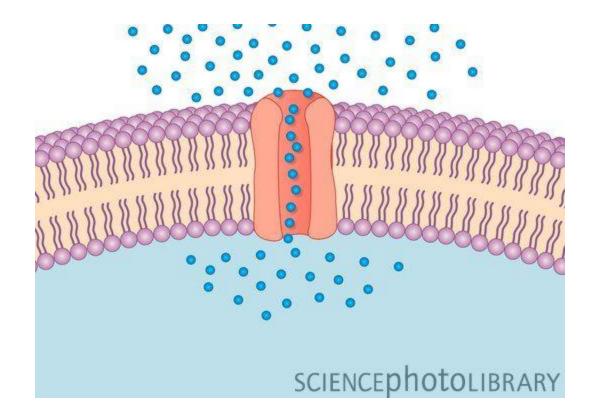
(a) Concentration gradients



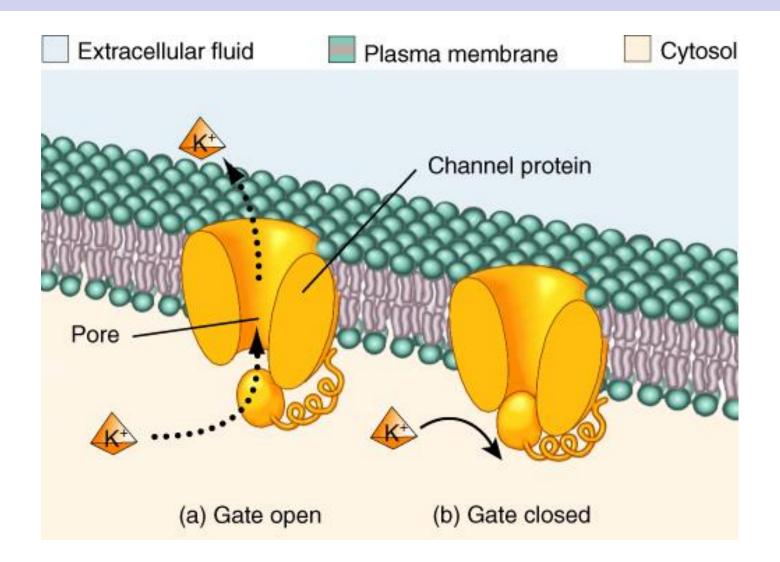
- CO2
- 02
- NO
- Steroid Hormones
- Monoglycerides

(b) Electrical gradient

Diffusion through Channels



Diffusion through Channels



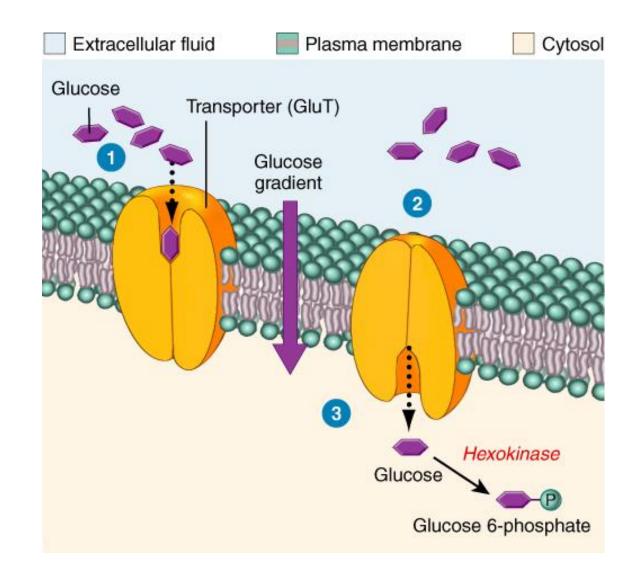
Fick's Law

- J = P.∆C
- $P = D.A/\Delta X$
- $J = D.A.\Delta C/\Delta X$

- J = Flux (Rate of diffusion)
- P = Permeability
- D = Diffusion Coefficient
- A = Surface area
- C = Concentration
- X = Membrane thickness

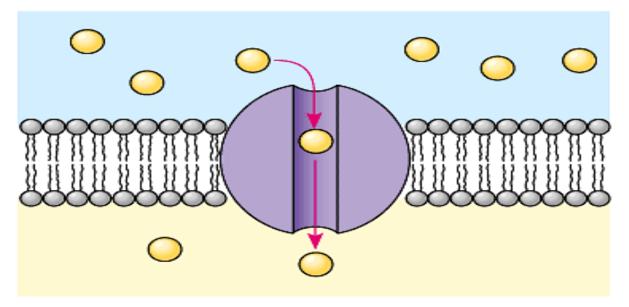
Facilitated Diffusion

Facilitated Diffusion

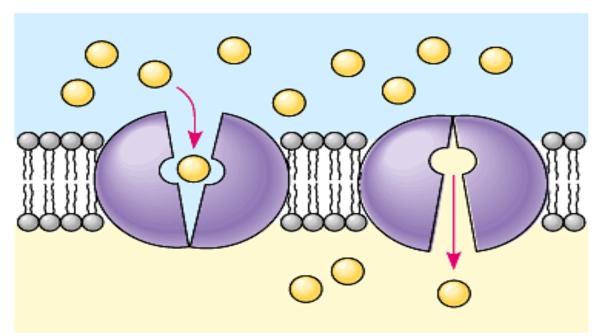


Facilitated Diffusion

- Aminoacids
- Glucose
- Galactose
- Fructose



(a)

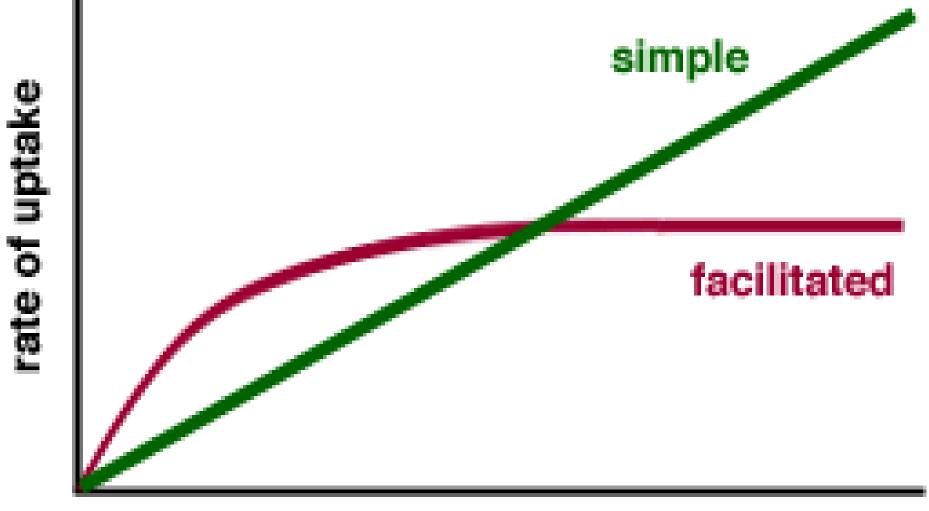


 Regarding the first Fick's law of diffusion,

Question

 By activation of channels, what parameter is changed?

Diffusion



concentration

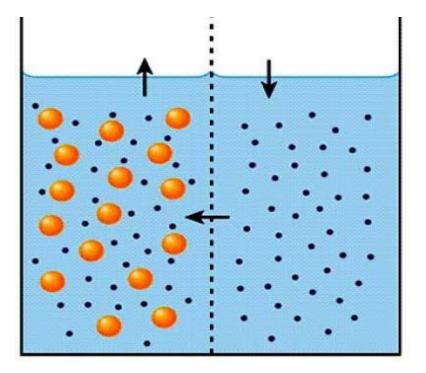
• Regarding **V max:** (maximum velocity of transport).

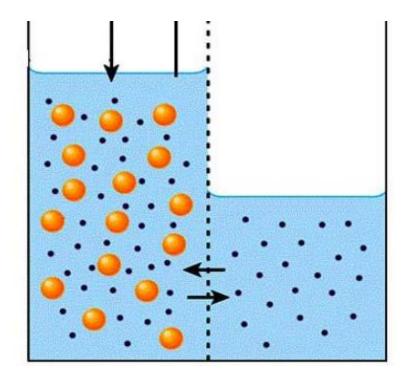
Questions

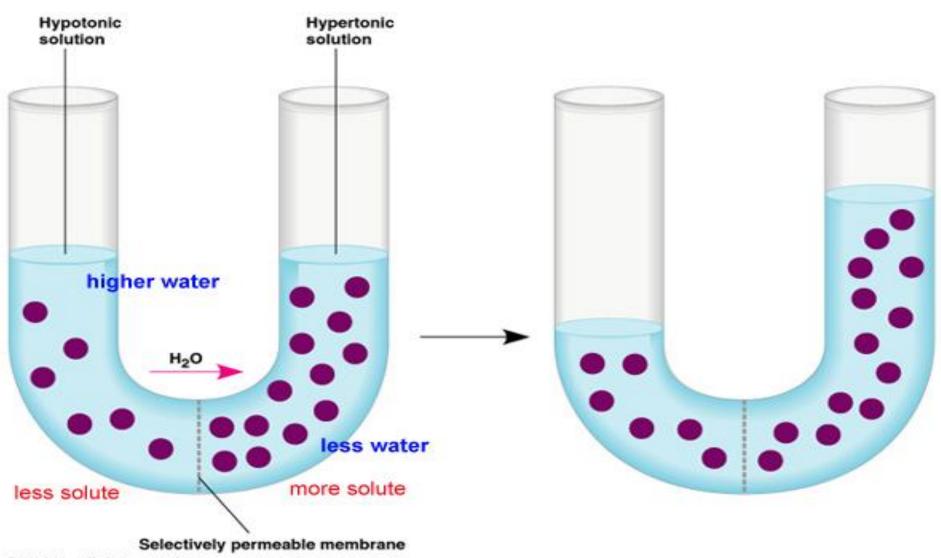
- Is it found in Simple diffusion or Facilitated diffusion?
- Explain why having this phenomenon?



Osmosis

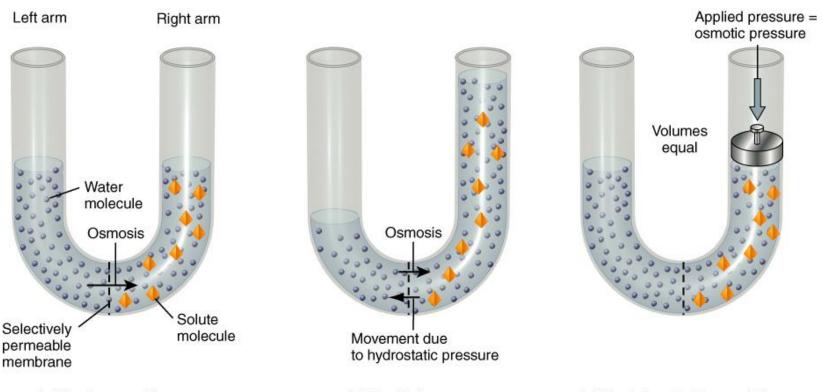






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Osmotic pressure



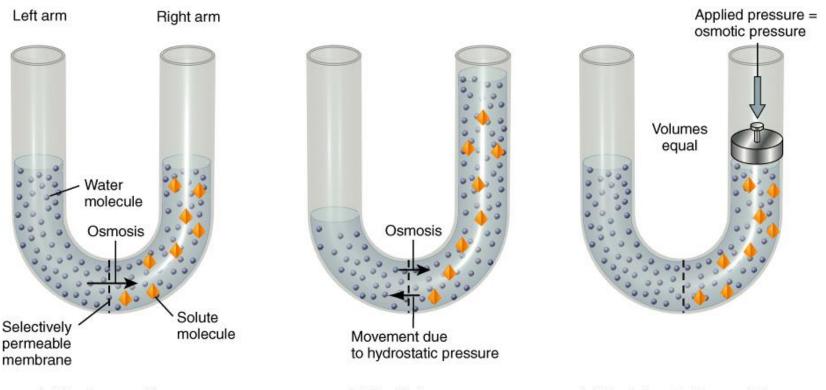
(a) Starting conditions

Van't Hoff's Law

$\pi = RTC$

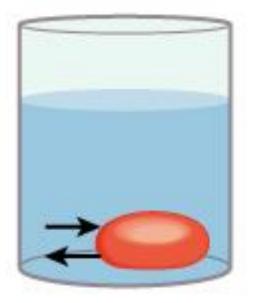
 π : osmotic Pressure R = Gas constant T = Absolute Temeprature C = Concentration

Osmosis vs. Filtration

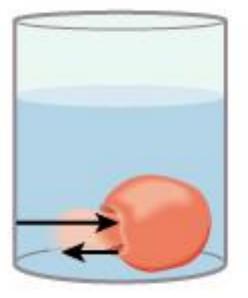


Tonicity of solution

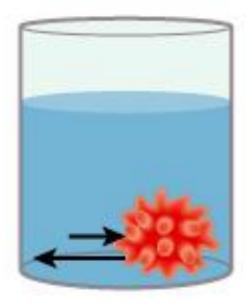
Isotonic solution



Hypotonic solution



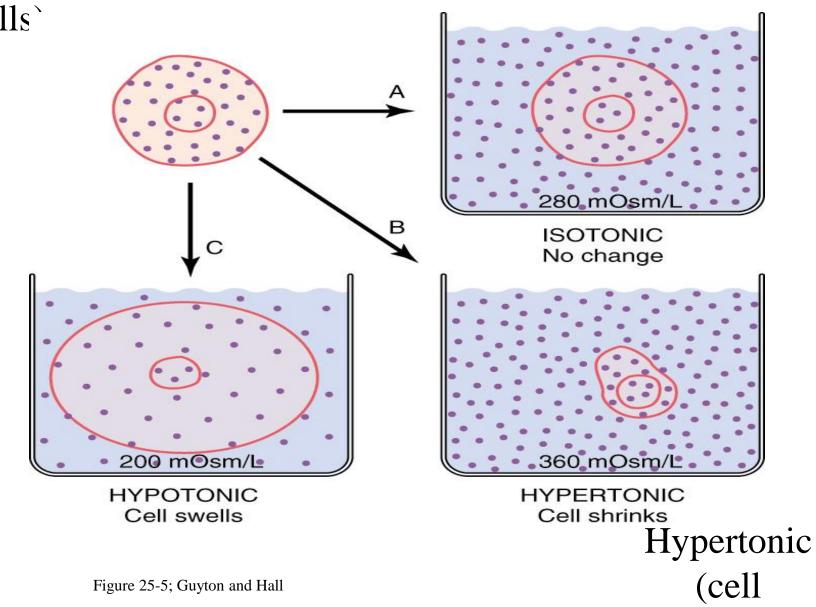
Hypertonic solution



(a) Normal RBC shape (b) RBC undergoes hemolysis (c) RBC undergoes crenation

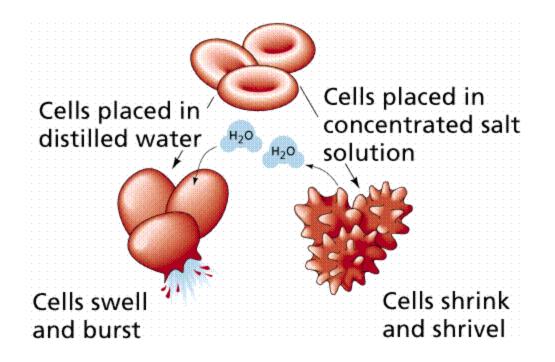
Hypotonic (cell swells`

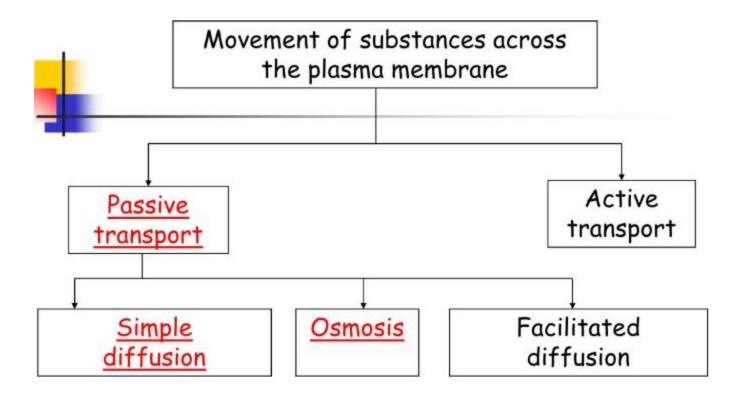
Isotonic (no change)



- - - \

What happens by placing Red Blood Cells in Hypertonic or hypotonic solution





Passive transport modalities

-Simple diffusion: transport through lipid bilayer, transport through channels, Ficks law of diffusion.

-Facilitated diffusion: by carriers

Differences in diffusion Kinetics between the previous modalities

Equivalent Concentration of particles.

Passive transport modalities

-**Osmosis**: concept of osmotic pressure (Van't Hoff's law), Oncotic (Colloid-osmotic) pressure. Osmolarity, Osmolality Hydrostatic pressure and filtration

Active transport modalities

 Primary active transport (ATP-ase carriers or Pumps) (functions of pumps: Na+/K+ pump, Ca++ pump, H+ pump, H+/K+ pump).

- Secondary active transport (Na+ dependent carriers) examples

Active transport modalities

- Vesicular transport: endocytosis, phagocytosis, transcytosis, pinocytosis and exocytosis and its control in secretory cells.-

TABLE 3.1 Passive Membrane Transport Processes

PROCESS	ENERGY SOURCE	DESCRIPTION	EXAMPLES
DIFFUSION			
Simple diffusion	Kinetic energy	Net movement of particles (ions, molecules, etc.) from an area of their higher concentration to an area of their lower concentration, that is, along their concentration gradient	Movement of fats, oxygen, carbon dioxide through the lipid portion of the membrane
Facilitated diffusion	Kinetic energy	Same as simple diffusion, but the diffusing substance is attached to a lipid-soluble membrane carrier protein or moves through a membrane channel	Movement of glucose and some ions into cells
Osmosis	Kinetic energy	Simple diffusion of water through a selectively permeable membrane	Movement of water into and out of cells directly through the lipid phase of the membrane or via membrane pores (aquaporins)
FILTRATION			
	Hydrostatic pressure	Movement of water and solutes through a semipermeable membrane (either through the plasma mem- brane or between cells) from a region of higher hydrostatic pressure to a region of lower hydrostatic pressure, that is, along a pressure gradient	Movement of water, nutrients, and gases through a capillary wall; formation of kidney filtrate

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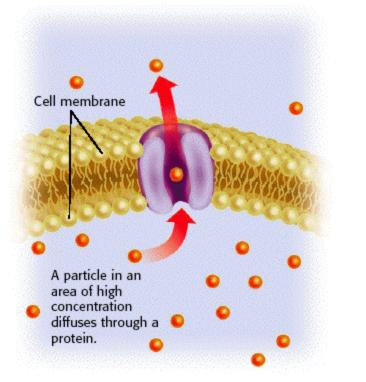
Active Transport Mechanisms

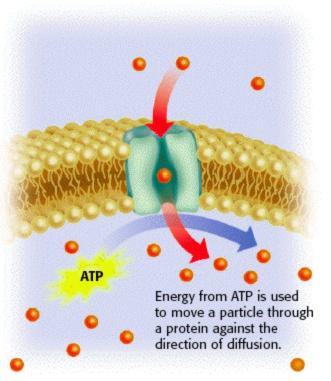
- Primary active transport
- Secondary active transport
- Vesicular transport

Passive and Active Transport

PASSIVE TRANSPORT

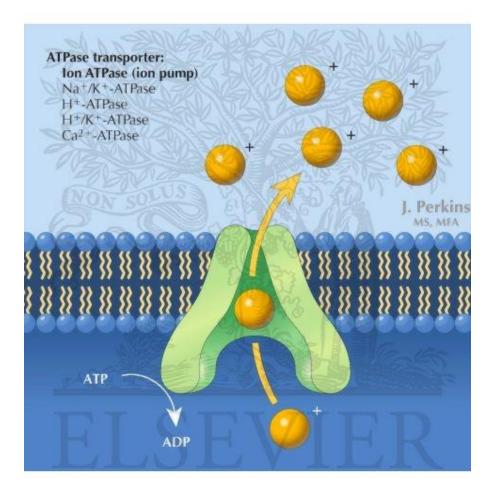
ACTIVE TRANSPORT

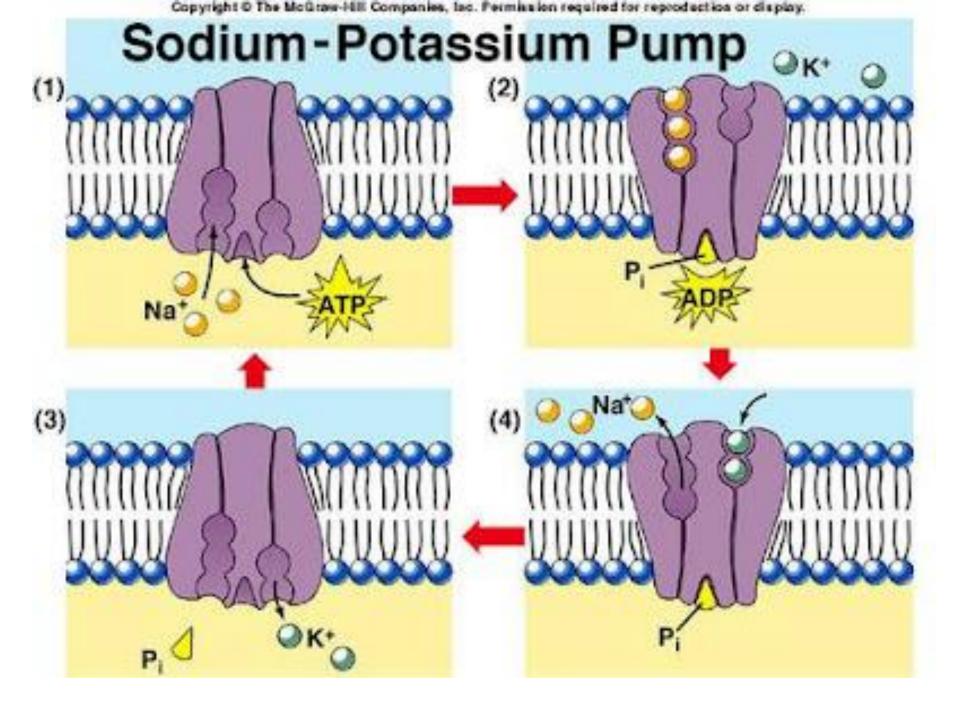


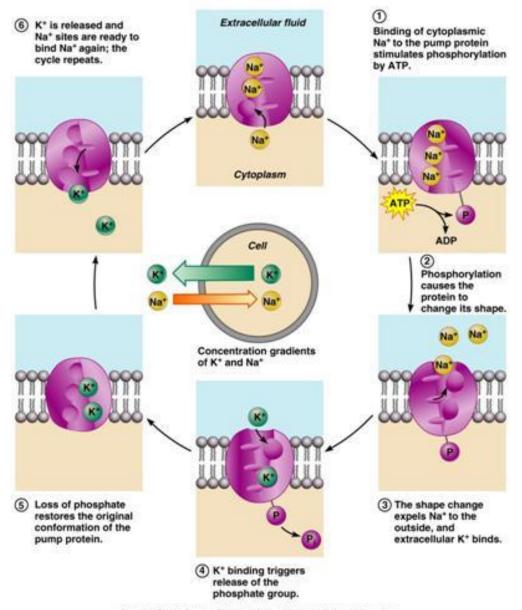


Types of Pumps

- Na+/K+ pump
- H+ pump
- H+/K+ pump
- Ca++ pump



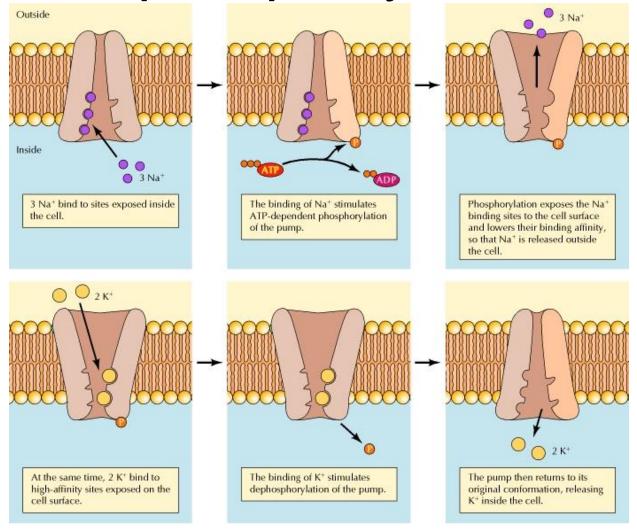




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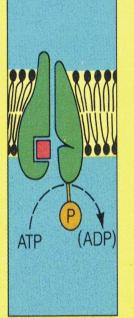
Figure 3.10

Phosphorylation & dephosphorylation



a Transport protein with two binding sites.

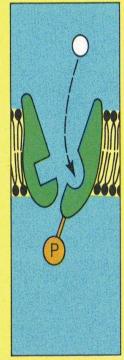
b Specific solute binds at one site.



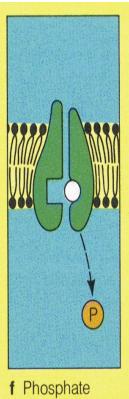
c Phosphate group is transferred from ATP to protein.

d Protein changes shape, pumps the solute across

membrane.



e The other binding site is now exposed, different solute binds to it.

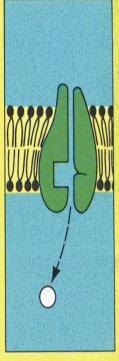


group is

released,

protein returns

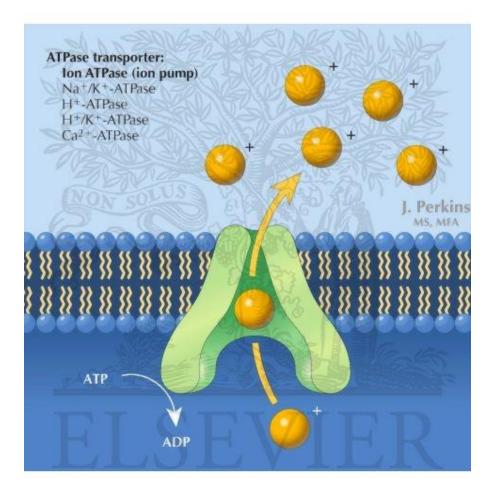
to original shape.



g The shape change causes the solute to be released.

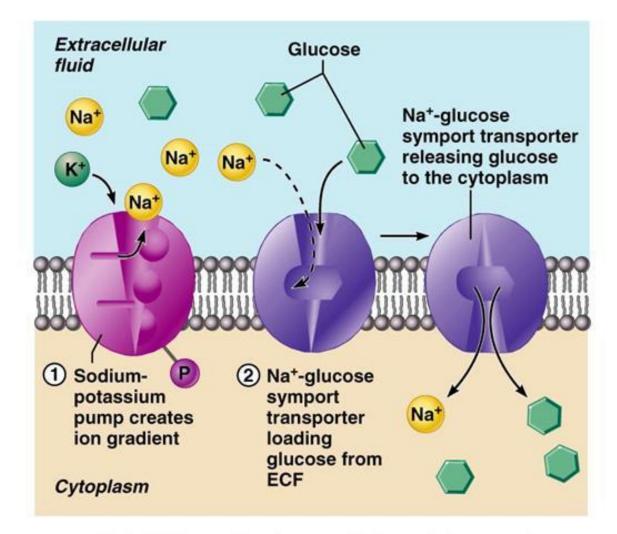
Types of Pumps

- Na+/K+ pump
- H+ pump
- H+/K+ pump
- Ca++ pump

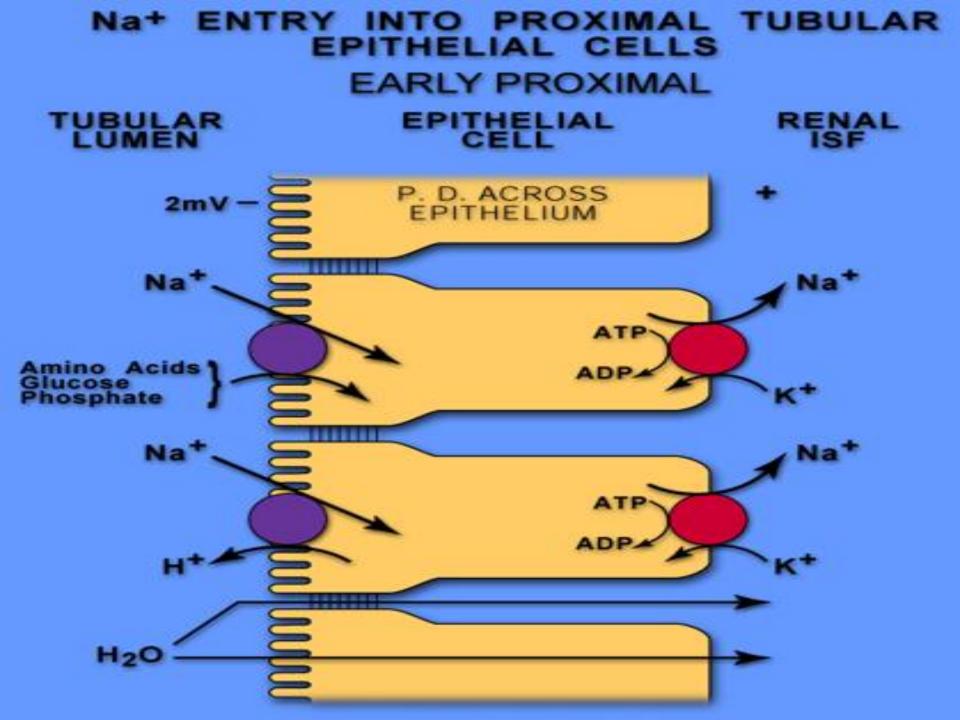


The importance of pumps for cell functions

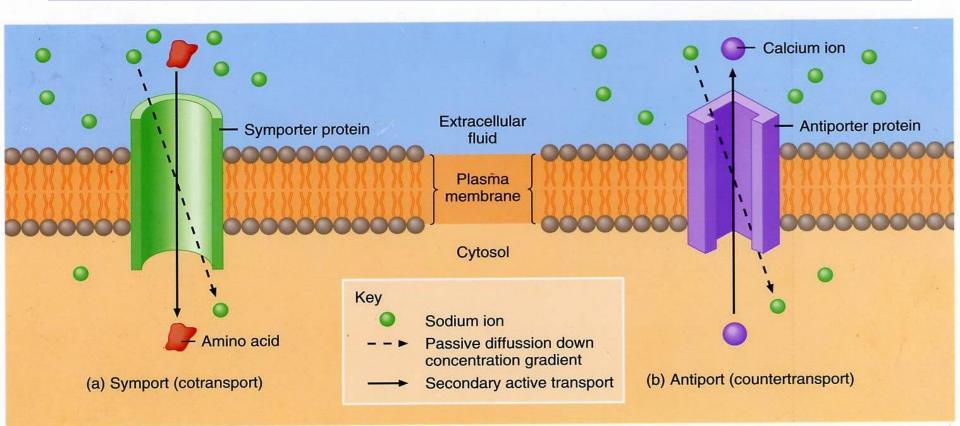
Secondary Active Transport



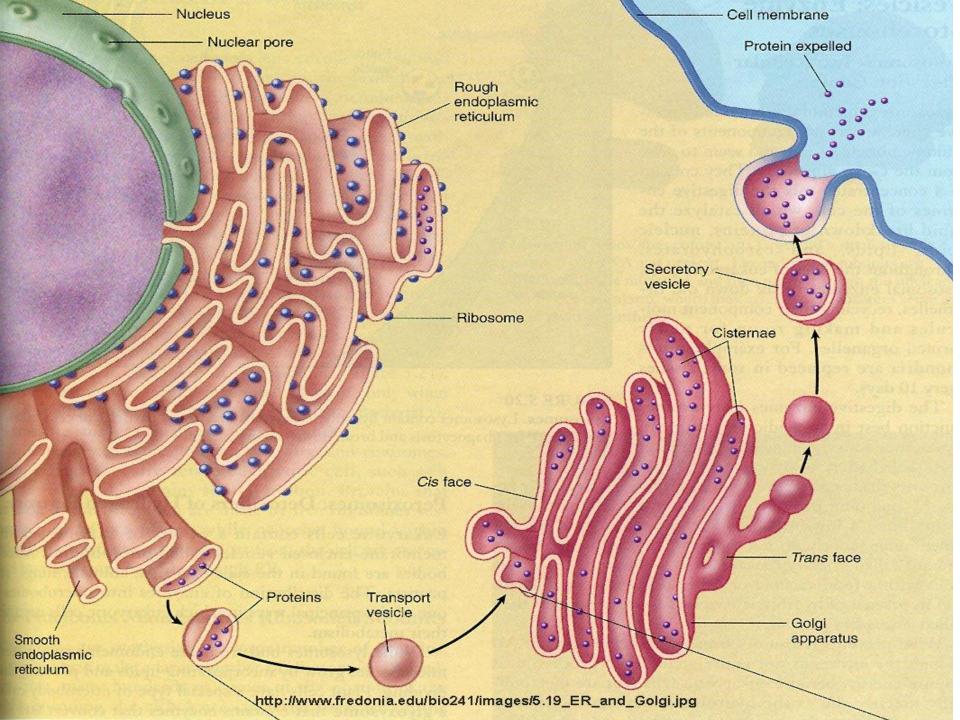
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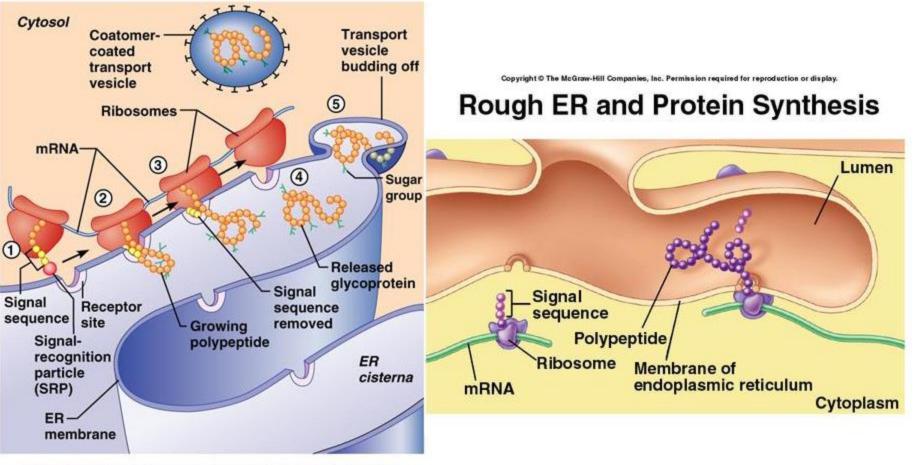
Secondary Active Transport -Co-transport -Counter transport



Vesicular Transport

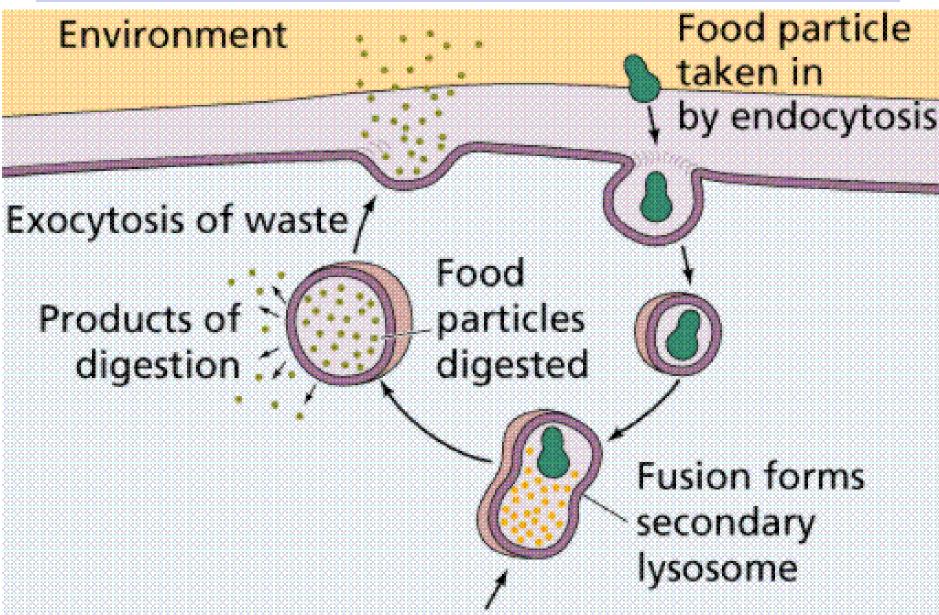


Formation of vesicles at ER

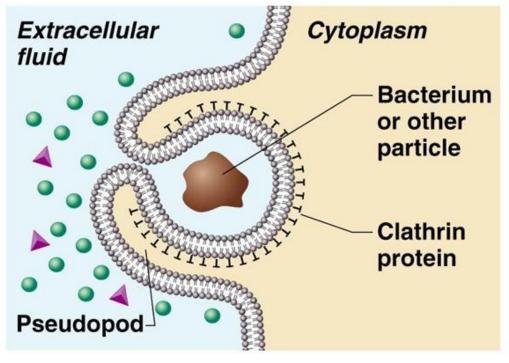


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Endocytosis



Phagocytosis



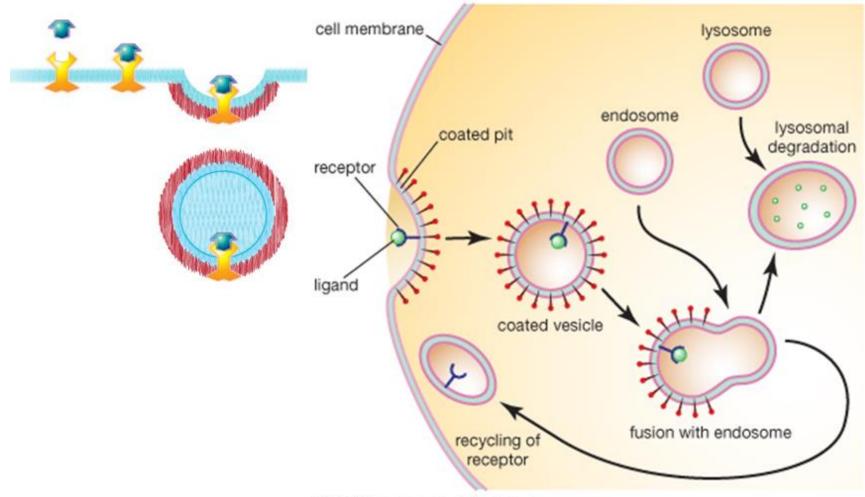
(b) Phagocytosis

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Figure 3.13b

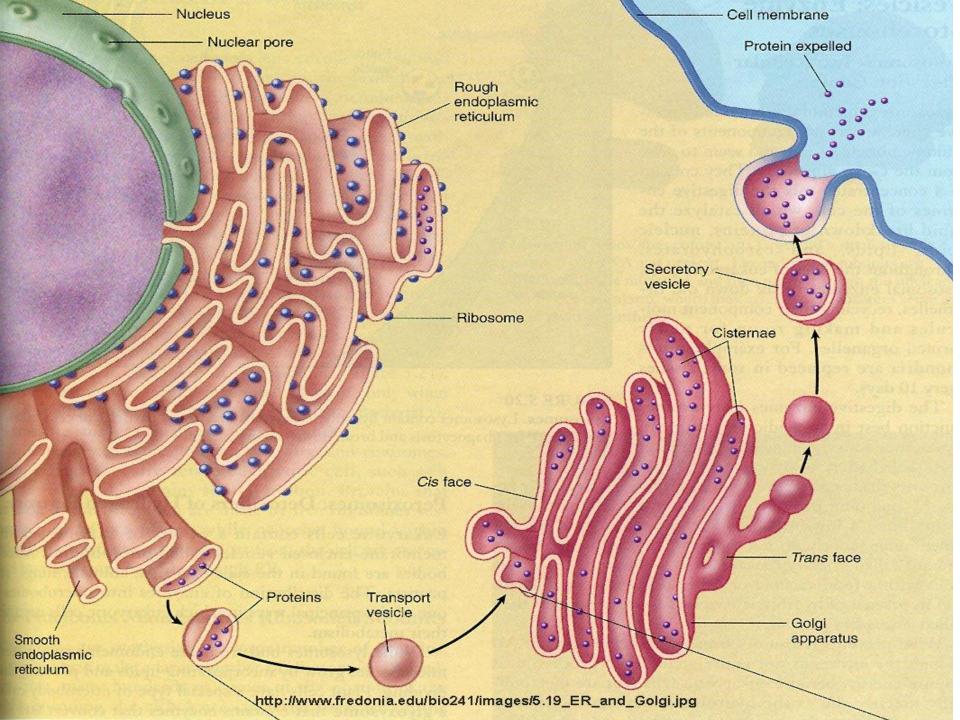
Receptor mediated endocytosis

Receptor-mediated endocytosis

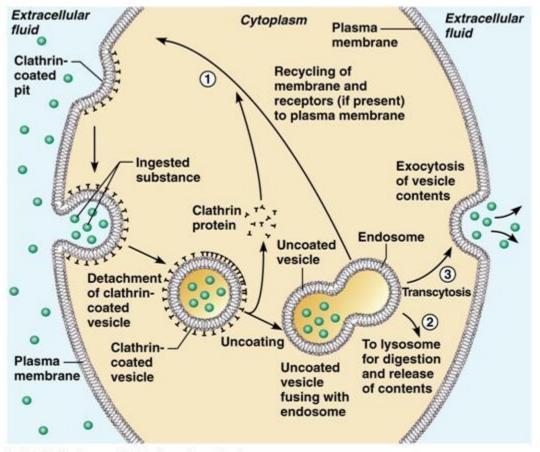


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Transcytosis



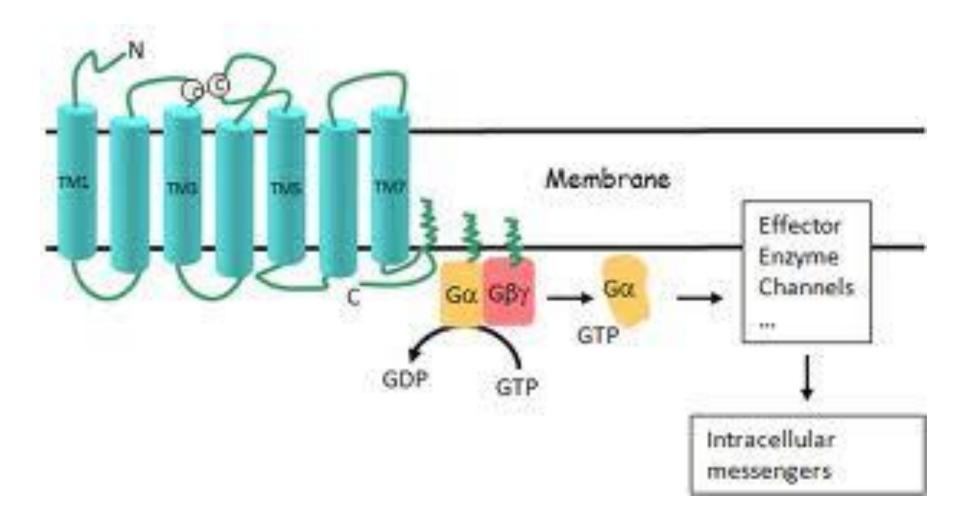
(a) Clathrin-mediated endocytosis

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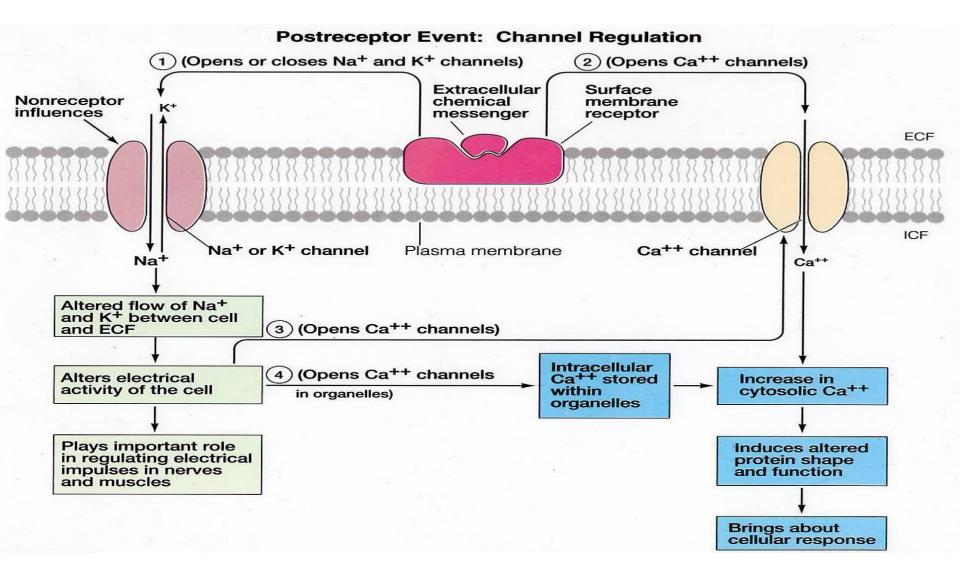
Transport Process	Description	Substances Transported
Osmosis	Movement of water molecules across a selectively permeable membrane from an area of higher water concentration to an area of lower water concentration.	Solvent: water in living systems.
Diffusion	Random mixing of molecules or ions due to their kinetic energy. A substance diffuses down a concentration gradient until it reaches equilibrium.	
Diffusion through the lipid bilayer	Passive diffusion of a substance through the lipid bilayer of the plasma membrane.	Nonpolar, hydrophobic solutes: oxygen, carbon dioxide, and nitrogen; fatty acids, steroids, and fat-soluble vitamins; glycerol, small alcohols; ammonia. Polar molecules: water and urea.
Diffusion through membrane channels	Passive diffusion of a substance down its electrochemical gradient through channels that span a lipid bilayer; some channels are gated.	Small inorganic solutes, mainly ions: K^+ , Cl^- , Na^+ , and Ca^{2+} . Water.
Facilitated Diffusion	Passive movement of a substance down its concentration gradient via transmembrane proteins that act as transporters; maximum diffusion rate is limited by number of available transporters.	Polar or charged solutes: glucose, fructose, galactose, and some vitamins.
Active Transport	Transport in which cell expends energy to move a substance across the membrane against its concentration gradient through transmembrane proteins that act as transporters; maximum transport rate is limited by number of available transporters.	Polar or charged solutes.
Primary active transport	Transport of a substance across the membrane against its concentration gradient by pumps; transmembrane proteins that use energy supplied by hydrolysis of ATP.	Na ⁺ , K ⁺ , Ca ²⁺ , H ⁺ , I ⁻ , CI ⁻ , and other ions.
Secondary active transport	Coupled transport of two substances across the membrane using energy supplied by a Na ⁺ or H ⁺ concentration gradient maintained by primary active transport pumps. Antiporters move Na ⁺ (or H ⁺) and another substance in opposite directions across the membrane; symporters move Na ⁺ (or H ⁺) and another substance in the same direction across the membrane.	Antiport: Ca ²⁺ , H ⁺ out of cells. Symport: glucose, amino acids into cells.
Transport In Vesicles	Movement of substances into or out of a cell in vesicles that bud from the plasma membrane; requires energy supplied by ATP.	
Endocytosis Receptor- mediated endocytosis	Movement of substances into a cell in vesicles. Ligand-receptor complexes trigger infolding of a clathrin-coated pit that forms a vesicle containing ligands.	Ligands: transferrin, low-density lipoproteins (LDLs), some vitamins, certain hormones, and antibodies.
Phagocytosis	"Cell eating"; movement of a solid particle into a cell after pseudopods engulf it to form a phagosome.	Bacteria, viruses, and aged or dead cells.
Pinocytosis	"Cell drinking"; movement of extracellular fluid into a cell by infolding of plasma membrane to form a pinocytic vesicle.	Solutes in extracellular fluid.
Exocytosis	Movement of substances out of a cell in secretory vesicles that fuse with the plasma membrane and release their contents into the extracellular fluid.	Neurotransmitters, hormones, and digestive enzymes.

Control of Transport and activity of Enzymes

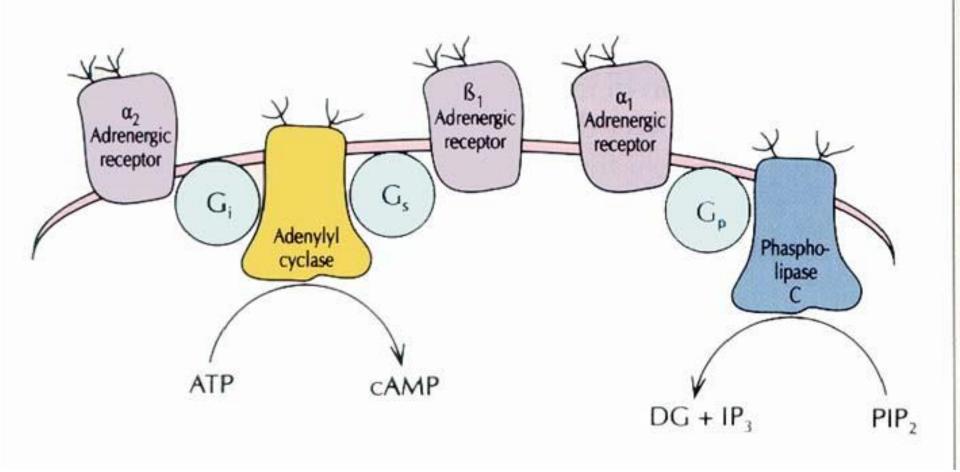
Receptors & Enzymes

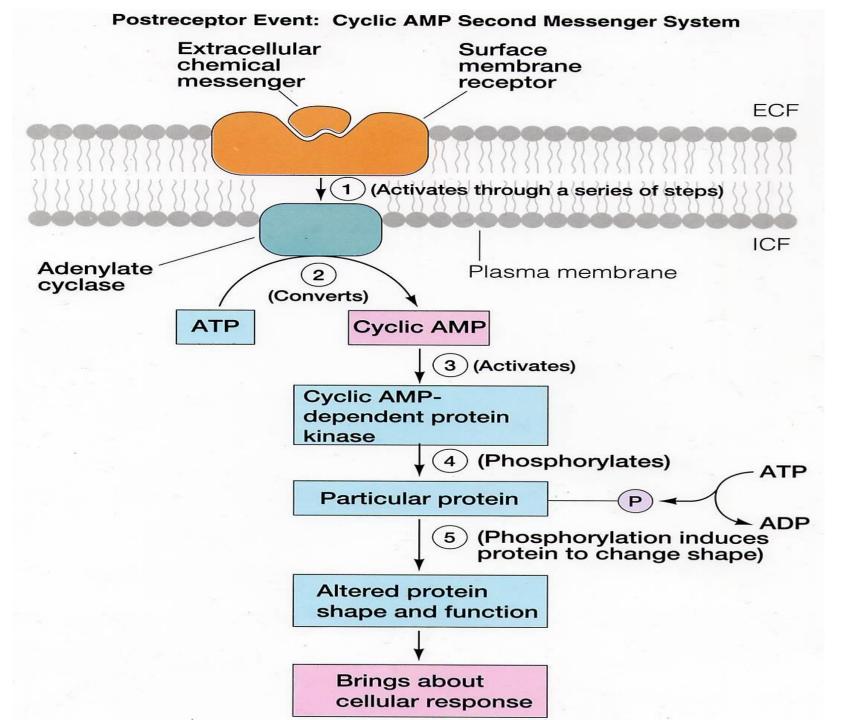


Receptors & Channels



Receptors & G proteins





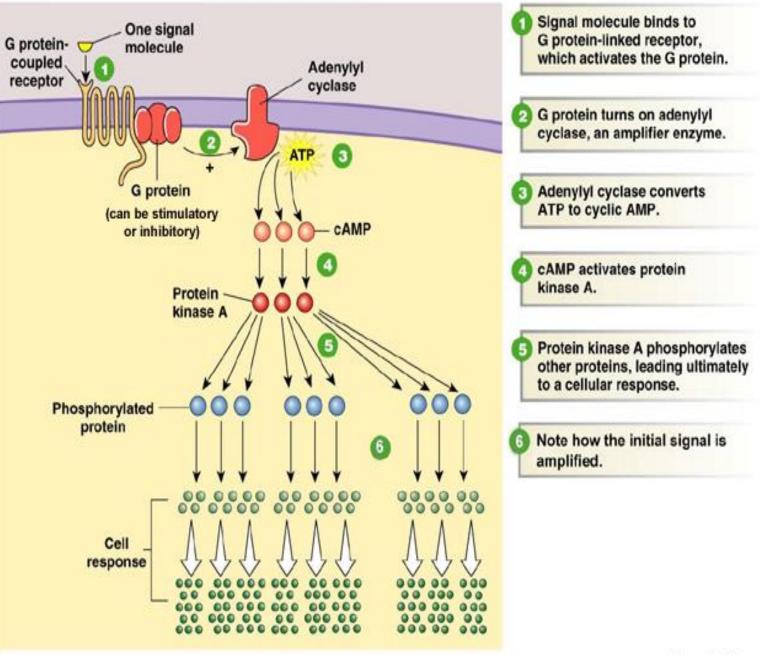
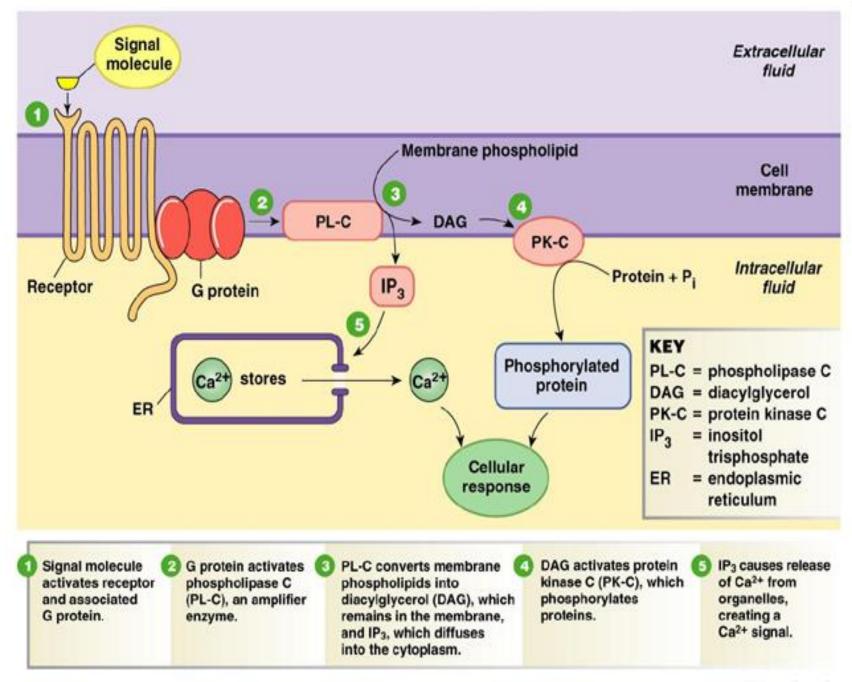


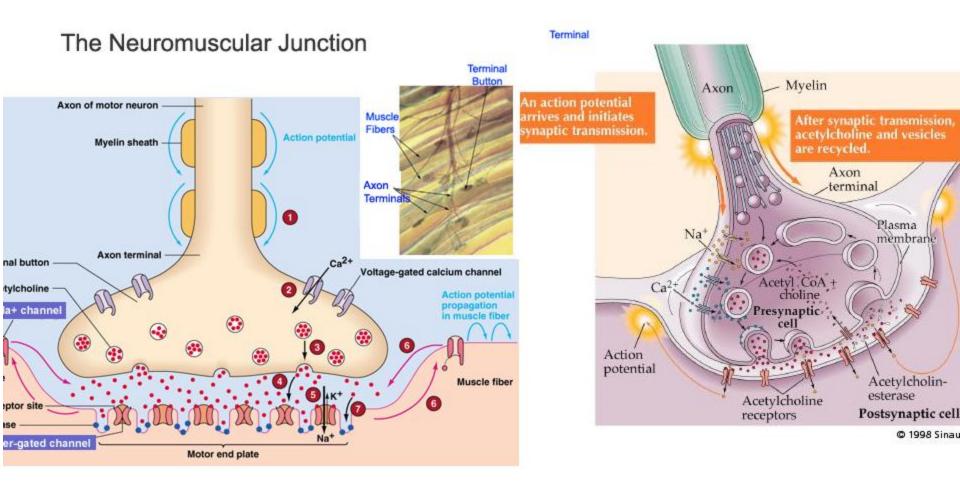
Fig. 6-11



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Fig. 6-12

Control of Exocytosis



Summary of Lectures

- -Homeostasis
- -Control of body functions by feed-back mechanisms to keep homeostasis
- -Functions of Cell organells (ER, Golgi complex, Mitochondria, Lisosomes)
- -Cytoskeletal structures and Functions:

Plasma membrane

- Membrane structure and function of **lipids of plasma membrane** including phospholipids, cholesterol and PIP2)

-Functions of Protein structures of plasma membranes (Channels, Carriers, Receptors, Pumps, enzymes, cell markers, G proteins, adhesion molecules)

-Activation of chemical gated channels

Passive transport modalities

-Simple diffusion: transport through lipid bilayer, transport through channels, Ficks law of diffusion.

-Facilitated diffusion: by carriers

Differences in diffusion Kinetics between the previous modalities

Equivalent Concentration of particles.

Passive transport modalities

-**Osmosis**: concept of osmotic pressure (Van't Hoff's law), Oncotic (Colloid-osmotic) pressure.Osmolarity, osmolality Hydrostatic pressure

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Active transport modalities

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