Biochemistry Mid Questions 2023

- 1. Which of the following is true about the isoelectric point of Arginine?
 - a) Is reached after adding 1.5 equivalents of a base
 - b) Is reached after the first step of titration
 - c) Equals approximately 10.5
 - d) Equals approximately 5.5

Answer: C

- 2. Which of the following is the reason that makes glycosaminoglycans (GAGs) act as shock absorbers?
 - a) The negative charge on the constituent monosaccharides
 - b) Their high concentration in the extracellular matrix

Answer: A

- 3. 100 mmol of an unknown amino acid at Ph=0.5 was completely titrated by adding 200 mmol of KOH, what is the unknown amino acid?
 - a) Glutamate
 - b) Methionine
 - c) Lysine
 - d) Cysteine

Answer: B

- 4. Phosphatidate is the backbone of?
 - a) Lecithin
 - b) galactocerebroside
 - c) Globoside
 - d) Glycerophospholipid

Answer: A

- 5. What is common between the plasmid of bacteria and mitochondrial DNA?
 - a) Both are circular
 - b) They make all the proteins in the cell
 - c) They can transfer among cells

Answer; A

- 6. Which of the following is the sequence of the complementary strand of the following DNA strand: ACGTAGCT ?
 - a) TGCATCGA

b) AGCTACGT

Answer: B

- 7. What are the main attraction forces which happen between nonpolar molecules?
 - a) Hydrophobic interactions
 - b) Van der Waals interactions

Answer: A

- 8. Why are Lecithin and bile acid great emulsifiers?
 - a) They are charged
 - b) They are amphoteric
 - c) They are amphipathic

Answer: C

- 9. Which of the following increases the fluidity of the plasma membrane?
 - a) The presence of unsaturated fatty acids
 - b) The presence of cholesterol

Answer: A

10. Which of the following is derived from Phenylalanine?

Answer: Tyrosine

11. Which of the following is the storage form of fatty acids?

Answer: Triacylglycerol

12. Which of the following share the same omega number?

Answer: Linoleate and arachidonate

13. The purine and pyrimidine are bound at......of the pentose sugar via.....linkage?

Answer: C1, glycosidic

- 14. Which of the following is the way of compensation for asthma?
 - a) Reabsorption of HCO3-ions
 - b) Inhibition of the production of ketone bodies

c) Excretion of HCO3- ions

Answer: A

15. Sorbitol can be obtained from which of the following?

- a) Oxidized glucose
- b) Reduced fructose

Answer: B

16. Which isn't true about Ka?

- a) Increasing Ka means stronger acid
- b) Strong acids have Ka
- c) We need 1 eq of OH- to reach the point of pKa in titration
- d) Ka is positively correlated with the concentration of H3O+ and OH-
- e) Weaker acids have higher pKa

Answer: C

17. What is the name of the following structure?

- a) Deoxyadenosine
- b) Deoxyadenylate
- c) Deoxycytidine

Answer: A



18. What is the relationship between alpha Galactose and beta Galactose?

- a) Epimers
- b) Anomers

Answer: B

19. What do triglycerides and glycerophospholipids share?

- a) 3 chains of fatty acids
- b) Both have charged groups
- c) The same backbone
- d) Both have phosphate

Answer: C

20. The pH in tomato juice is 4.5, what is the [OH-]?

Answer: 10^-9.5

21. Serotonin and Melatonin are derivatives of?

Answer: Tryptophan

22. Which group contains only hydrophilic amino acids?

Answer: Tyr, Ser, Thr

23. Cholesterol is the precursor of all the following except?

Answer: Thromboxane

- 24. Which of the following has an N-glycosidic bond?
 - a) ATP
 - b) N-acetylglucosamine

Answer: A

25. Glycogen and Amylose differ in?

Answer: the presence of branching

26. Which of the following has a secondary amine?

Answer: Proline

27. What is a characteristic of water?

Answer: Amphoteric

28. Which is the most efficient buffer?

Answer: Bicarbonate buffer

29. Which of the following is true regarding wax?

Answer: Long hydrocarbon chain

30. In a solution of 0.25 M NaH2PO4 and 0.35 M Na2HPO4, 0.05 M of HCl was added, calculate the pH. (pKa is 7.2)

Answer: 7.2

31. The immediate precursor of eicosanoids?

Answer: Arachidonic acid

- 32. A hypothetical amino acid with a positively charged R group and pKa=12, what is the charge at pH=4?
 - a) +2
 - b) +1

c) 0

Answer: B

- 33. Titration of 40ml H3PO4 and has a Molarity of 0.3, with 150ml of KOH, what is the Molarity of KOH?
 - a) 8M
 - b) 0.08M
 - c) 80M
 - d) We can't titrate a weak acid with a strong base.

Answer: B