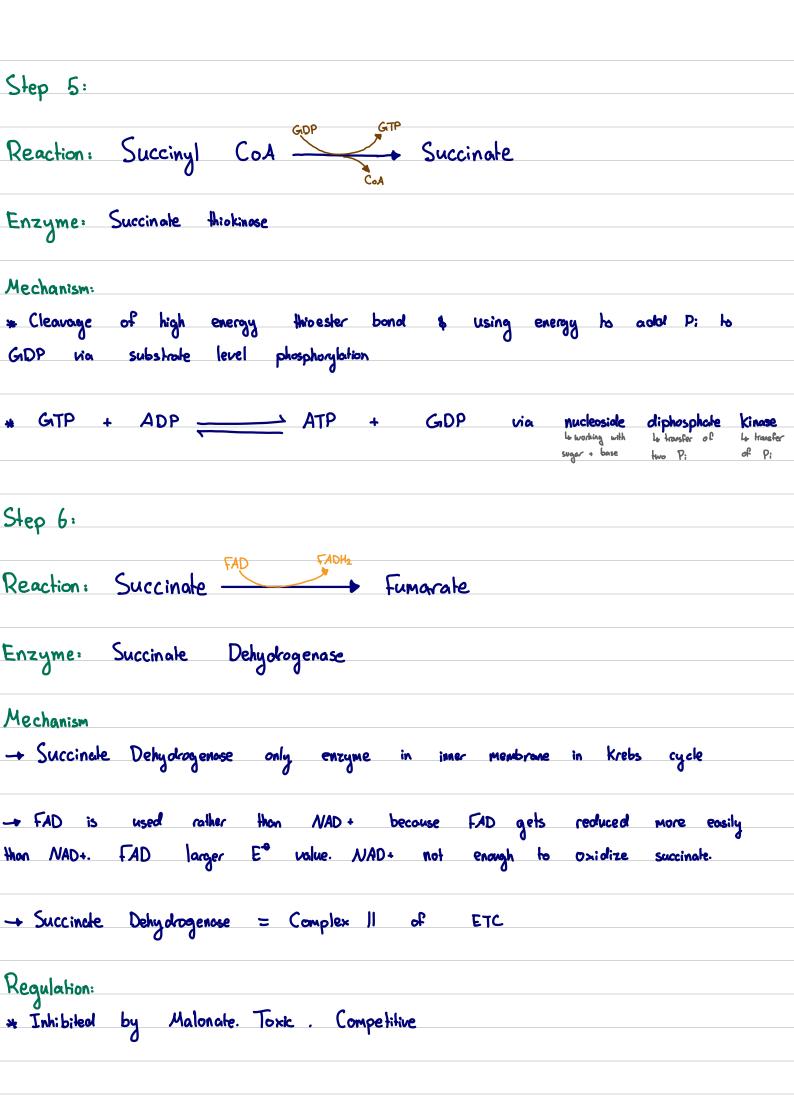


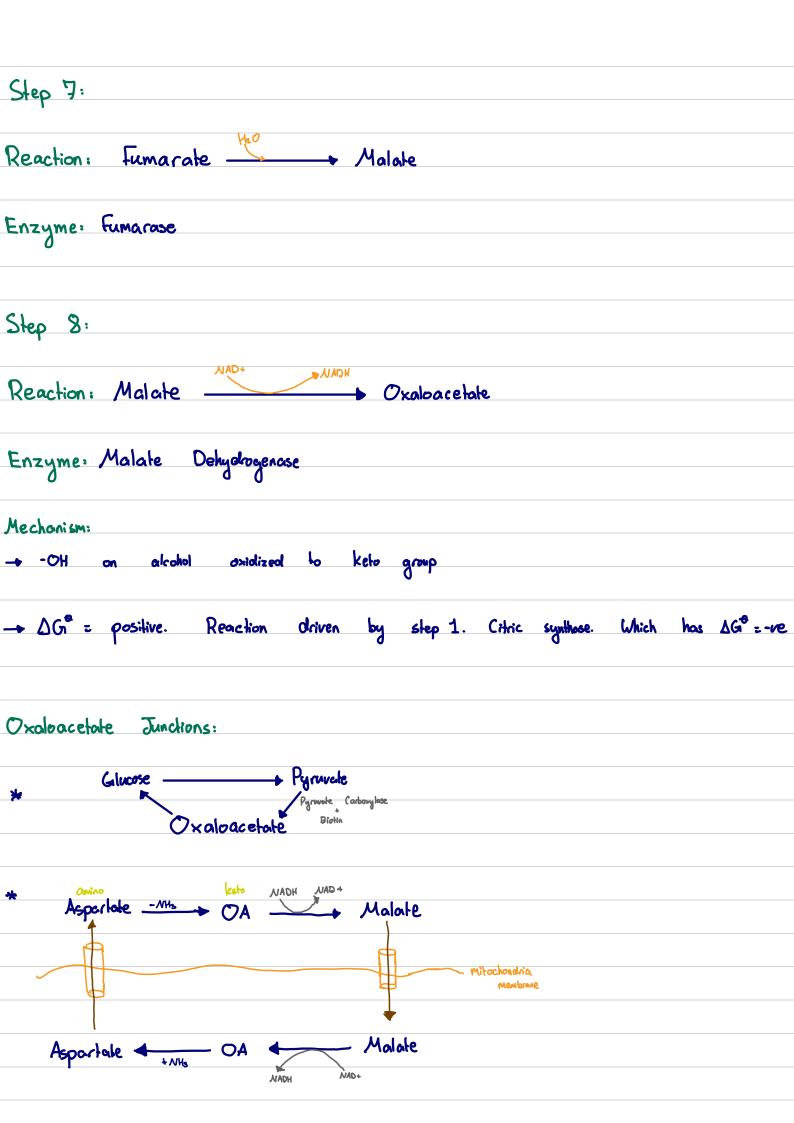
FAD vs NAD: ()→FAD gains e one by one, sequential. By different sources. form of H. \rightarrow NAD gains pair of electrons at once by same source. Form of hydrole \ddot{H} $\bigcirc \rightarrow FAD$ used in succinate dehydrogenose. + FAD used in a-keto glutarate dehydrogenase complex, FAD oxidized disulfide bridge in Lipoic acial in Transacylase to form FADH2 then transfers electrons to NAD+ - NAD used in most dehydrogenases. Oxidizes alcohol to ketones 3 - Since FAD forms free radicals, if forms fight Covalent bonds to enzymes. This changes the E^O - NAD can be free () -> NADH used in regulating krebs cycle, acting as inhibitor. While FADH2 not used $\bigcirc \rightarrow NADP^+$ used to convert malate to pyruvate Malate Malate Pyruvale Matate MADPH ATP ADP + Pi Cos Oxa loace tale Malate NADH NAD+

Details of steps: Step 1: Reaction: Oxaloacetate + Acetyl CoA - Citrate + CoASH Enzyme: Citrate synthase Regulation: Activation / Inhibition of enzyme - Activated by oxaloacetate which results in Conformational shape change allowing Acelyl - CoA to bind to enzyme. - Inhibited by citrate - Inhibited by ATP allostenically - Inhibited by NADH - Inhibited by succinyl-CoA Other regulations using citrate: - Inhibiting rate -limiting step of glycolysis. Inhibiting phosphofructokinose - In gluconeogenisis, it stimulates fructose-1,6-Bisphosphatase - Plays role in fatty acid synthesis La Provides Acetyl CoA source Lo Activates a cetyl CoA carbonylase Lo Inside mitochandria, fatty acid andized. Outside mitochandria fatty acid synthesized Le Citrate breaks down to Acetyl COA + Oxaloacetate. Oxaloacetate reduced to malate to cross mitochandrial membrane then axidized actain to avaloacetate

Step 2: Reaction: Citrate ____ Cis-aconitate ____ Iso-citrate 3° alcohol 2° alcohol - isomerization Enzyme: Aconilase Regulation: - Inhibited by Fluoroacetate. Toxic, stops Krebs cycle. Non-competitive - Aconitase enzyme contains Fe-S clusters - isomerization RXN Step 3: NAD+ ← ∝-ketoglularate Reaction: Isocitrate Enzyme: Isocitrate dehydrogenose Regulation: -> Dehydration + decarboxylation. - Rate - Limiting Step !! Best regulation - Inhibited by NADH - Inhibited by ATP - Activated albeterically by ADP C a²⁴ => shift to left. Lower Km. Higher affinity - Activated allosterically by lo muscle - contraction => octive => more ATP

Step 4: NADH NAD+ Reaction: oc- Keto glutarate + CoA Succinyl COA Enzyme: a-ketoglutarate complex Mechanism of a-keto acid dehydrogenese complexes: -> Include of-ketoglutarate / pyruvate / branched chain keto acid DH complexes le alonine keto acid - Three enzymes. 5 cofactors - Take Louing Care For Mancy. TLCFN 1) Decarboxylase (E1/E2/E3). Cofacturs: Thiamin Pyrophosphate TPP (2) Transacylose. Adds CoA and uses -S-S- to transfer H to FAD. Cofactors: Lipoic acid 3 Deby drogencise. FAD takes M from Lipoic acid on transacylase then it transfers them to NAD". NADM formed. - Me chanism: 1: CO2 removed from a-KG. TPP used. Decarboxylase 2: -S-S- Oxidize a-kG to form -SH + -SH. Lipoic aciel used. Transacylase 3: COA is added to Succinyl COA released. Transacylase 4: FAD takes H From -SH & Forms -S-S- again 5: FAD transfers H to NAD+ Forming NADH 6: Energy conserved in NADH & thioester of COA Regulation: - Activated by Ca2+ - Inhibiled by NADH - Inhibited by GIP - Inhibiled by Succinyl - CoA - Inhibited by Arsenite. Toxic. Non-competitive





Toxins on the Krebs cycle: * Flouroacetale: Aconitase non-competitive inhibitor Citrate — Isocitrate * Arsenite: a-keto glutarate olehydrogenase non-competitive inhibitor a-keto glutarate COA COL * Malonate: Succinate Dehydrogenase competitive inhibitor Succinate FADH FADH Krebs Cycle intermediates used in other pathways: * Citrate: Lo fatty acid synthesis in liver 4 Activales Acetyl CoA carbonylose 4 Breaks down to form Acetyl CoA, building blood of fatty acids. 4 Occurs in cytosol NOT mitochandra * X-Ketoglutarate: Lo Converted to glutamate 4 Forming GABA Lo Neurotransmitter on NS * X- Ketoglularale: La Converted to glutamine La skeletal muscle i other tissues Le Protein synthesis

* Succinyl - CoA: Lo Heme synthesis in bone marrow * Malates 4 Increase blood glucose when fashing vion gluconeogenisis L+ In liver * Oxaloacetate: 4 Amino acid synthesis Anaplerotic Reactions: - Since intermediates can be used, we need reactions that replenish those intermediates called anaplerotic reactions. * Aspartate ---- Oxalo acetate * Alanine ----- Pyruvate + HCO3 ----- Oxaloacetate !! imp * Glutamate - X-Ketoglutarate * Amino Acids - Propionyl CoA - Succinyl CoA * Amino Acids ----- Fumarate -+ All anapleratic reactions use amino acid except pyruvale to oxaloacetale

